



Digester and Recaust Automatic Liquor Titrator Measurement Technology

Presented by

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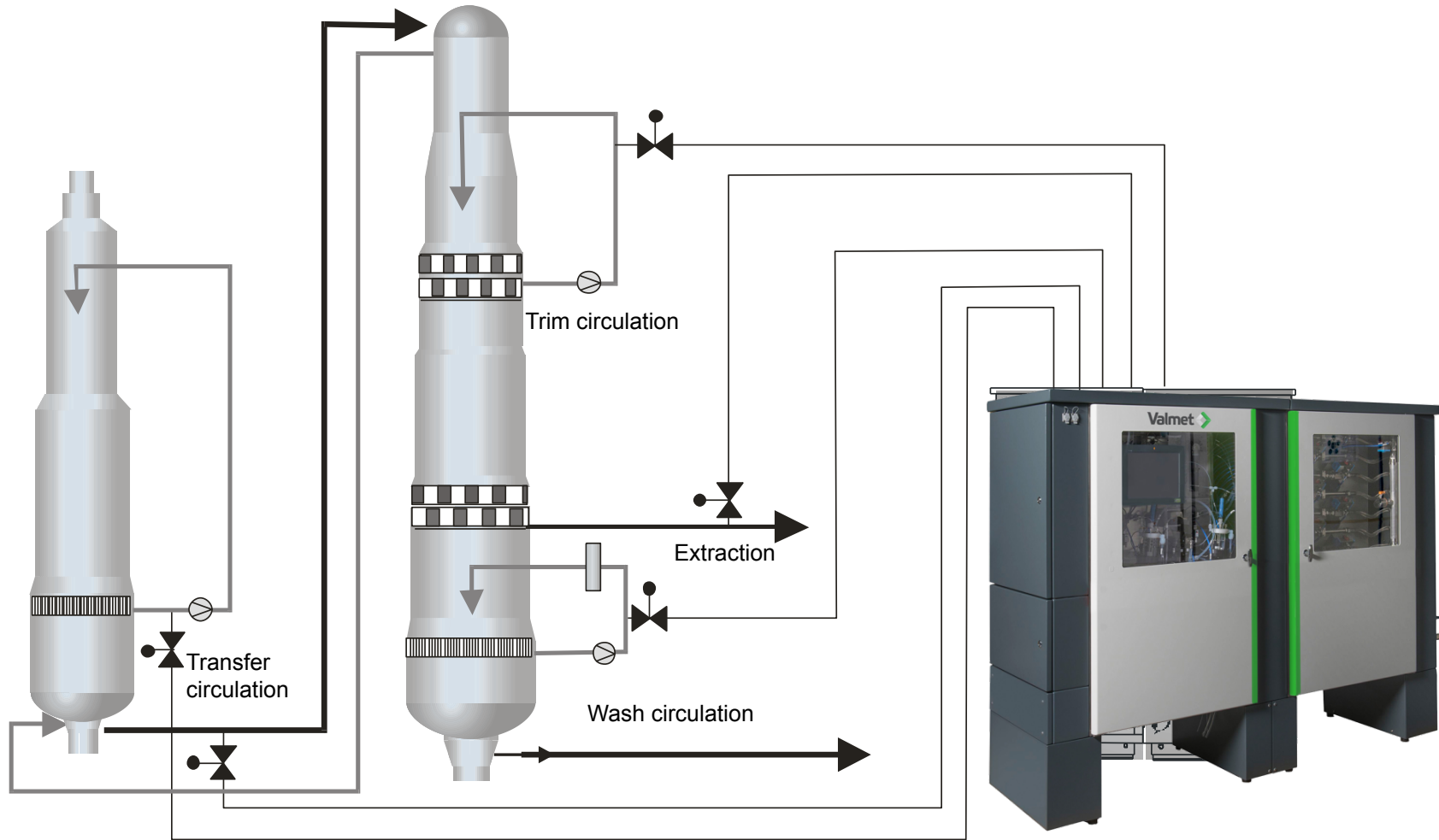
Valmet Solutions Manager

Cooking Liquor Titrator

- Standard analysis method for black liquor residual alkali
 - EA
- Complete white liquor analysis as an optional feature
 - NaOH, Na₂CO₃, Na₂S
 - Effective alkali (EA), Active alkali (AA), Total Titratable Alkali (TTA), Causticizing Efficiency (CE %), Sulphidity (S%)
- Analysis based on an inflection point titration that eliminates the effect of any pH electrode drift
- Sample Handling unit and Analysis unit can be separated to facilitate installation if needed.
 - There are some distance requirements that need to be considered if this option is used.



Typical sampling points for a continuous digester



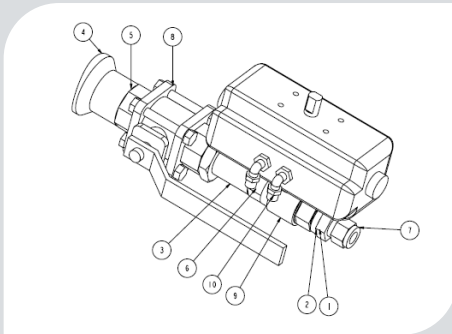
Process Samplers

Sample line:

- Maximum: 16 lines
- AISI 316L

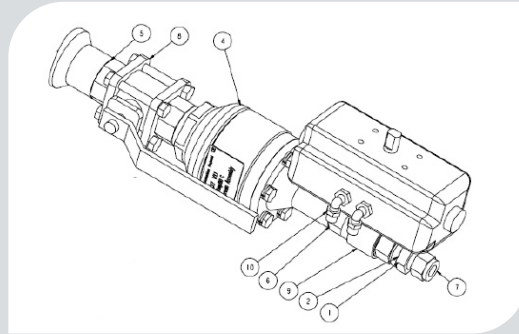
Samplers:

- Installation with a coupling welded to the process pipeline



Sampler SD 711

- White liquor or shive-free black liquor
- Sample line max. 100 m



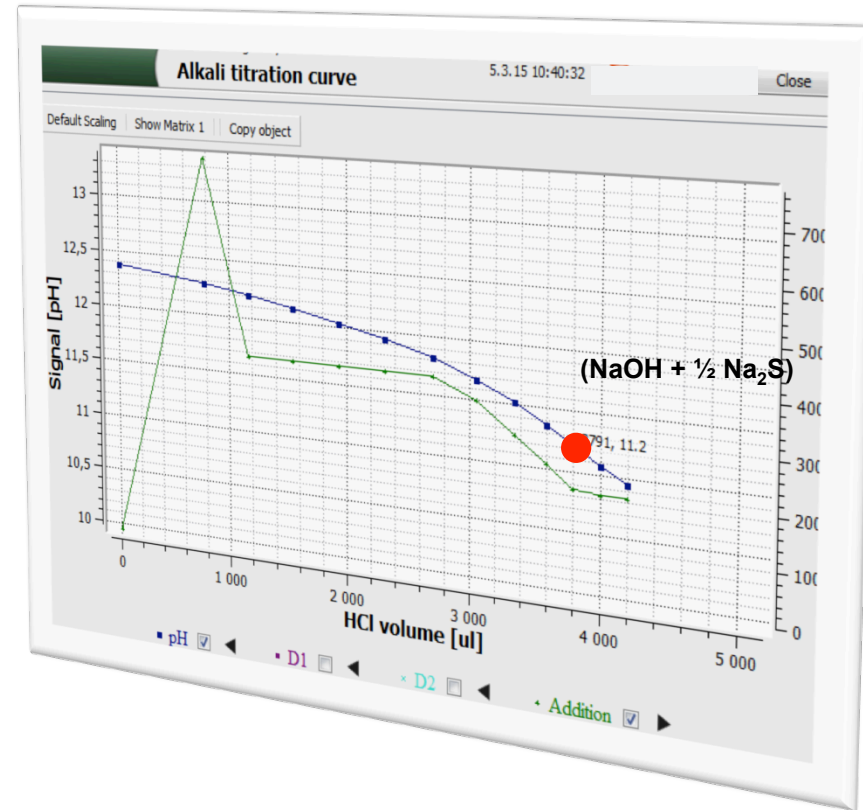
Sampler SD 712

- Shive-containing liquors
- Sample line max. 30 m

Analysis method

Black liquor

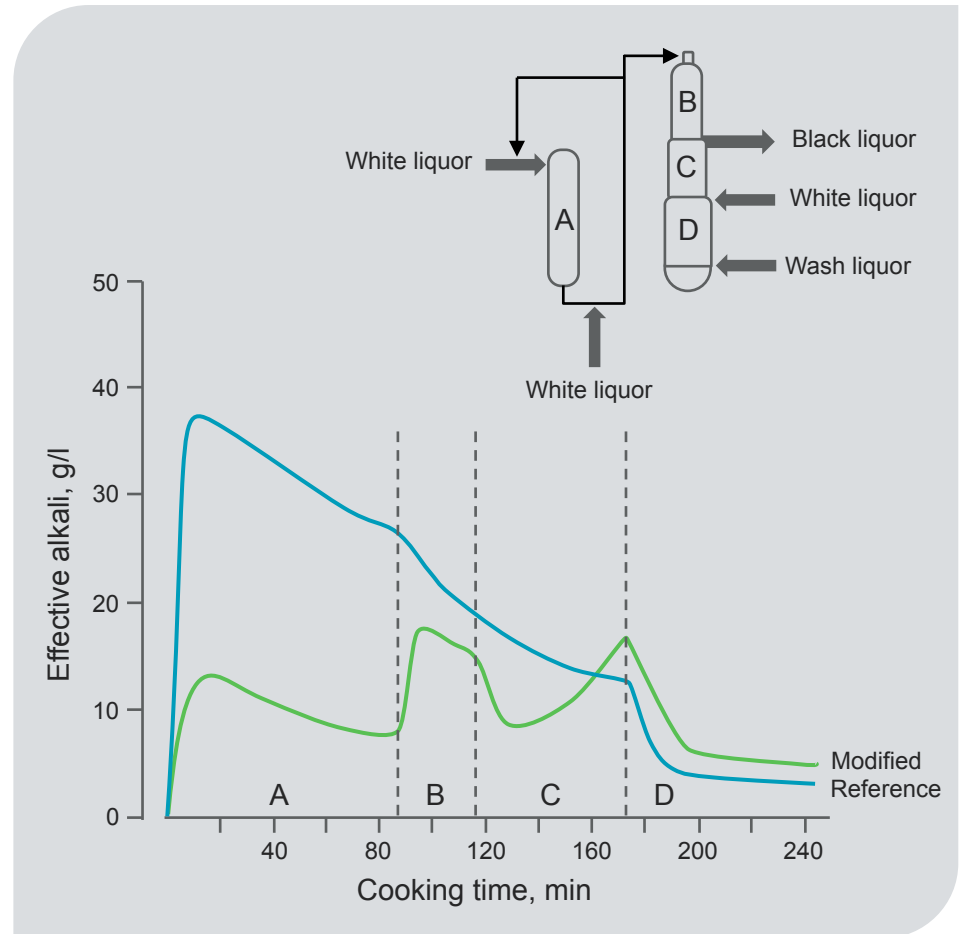
- SCAN N 33:94
- Potentiometric titration with hydrochloric acid (HCl)
- Sodium carbonate is added to give a better inflection and buffer the titration solution
- Autocalibration feature for adaptation to changing process conditions
- Possible to use pH end point titration if desired.
 - Requires regular pH probe calibration



Managing the alkali profile

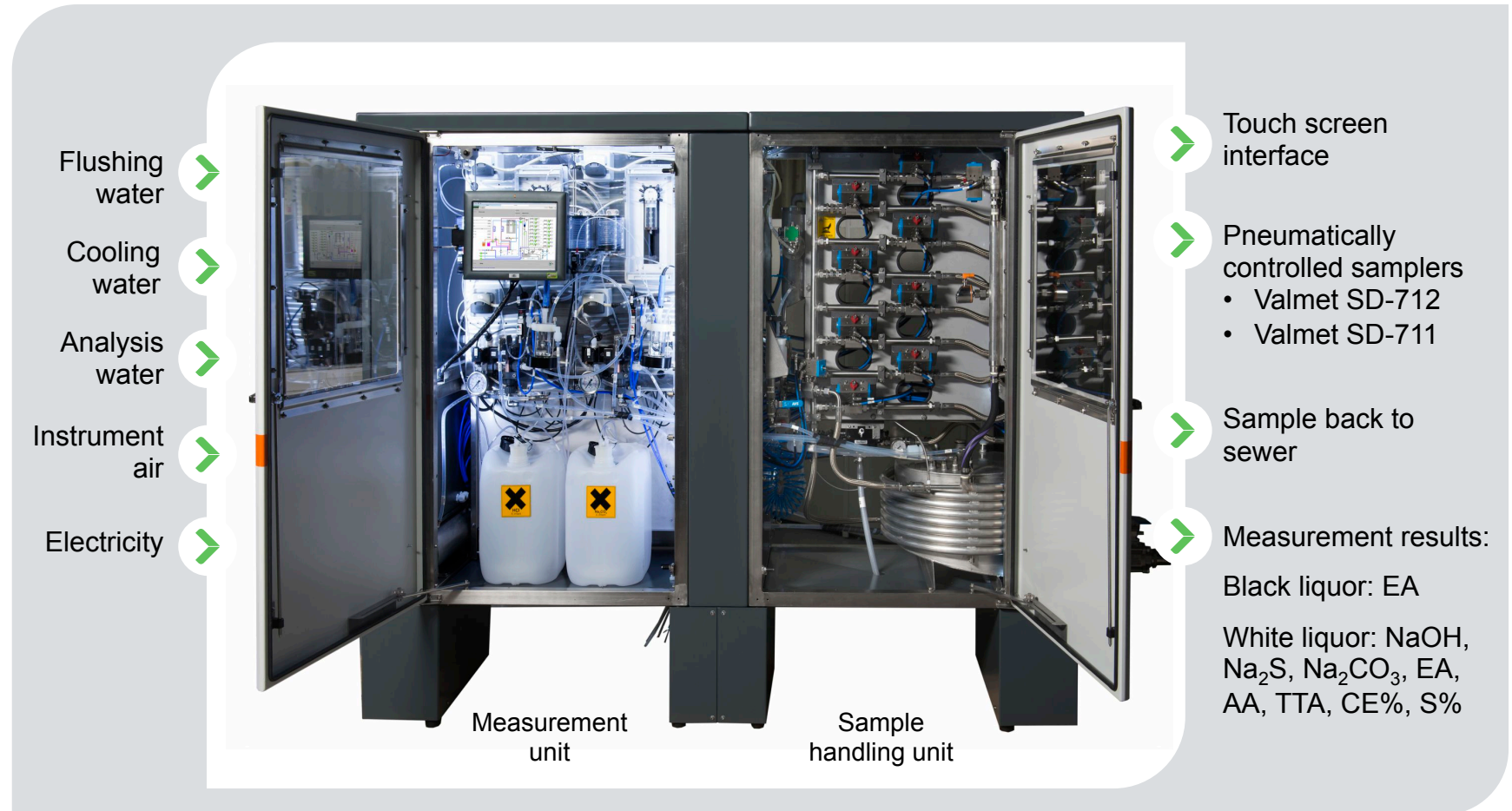
Reliable and dependable measurement information

- Automatic sampling and analysis for real liquor chemistry results
- Absolute measurements
- Managing the alkali charge and profile
- System point of view for operators
- Seamless integration into process and mill
- High frequency measurements



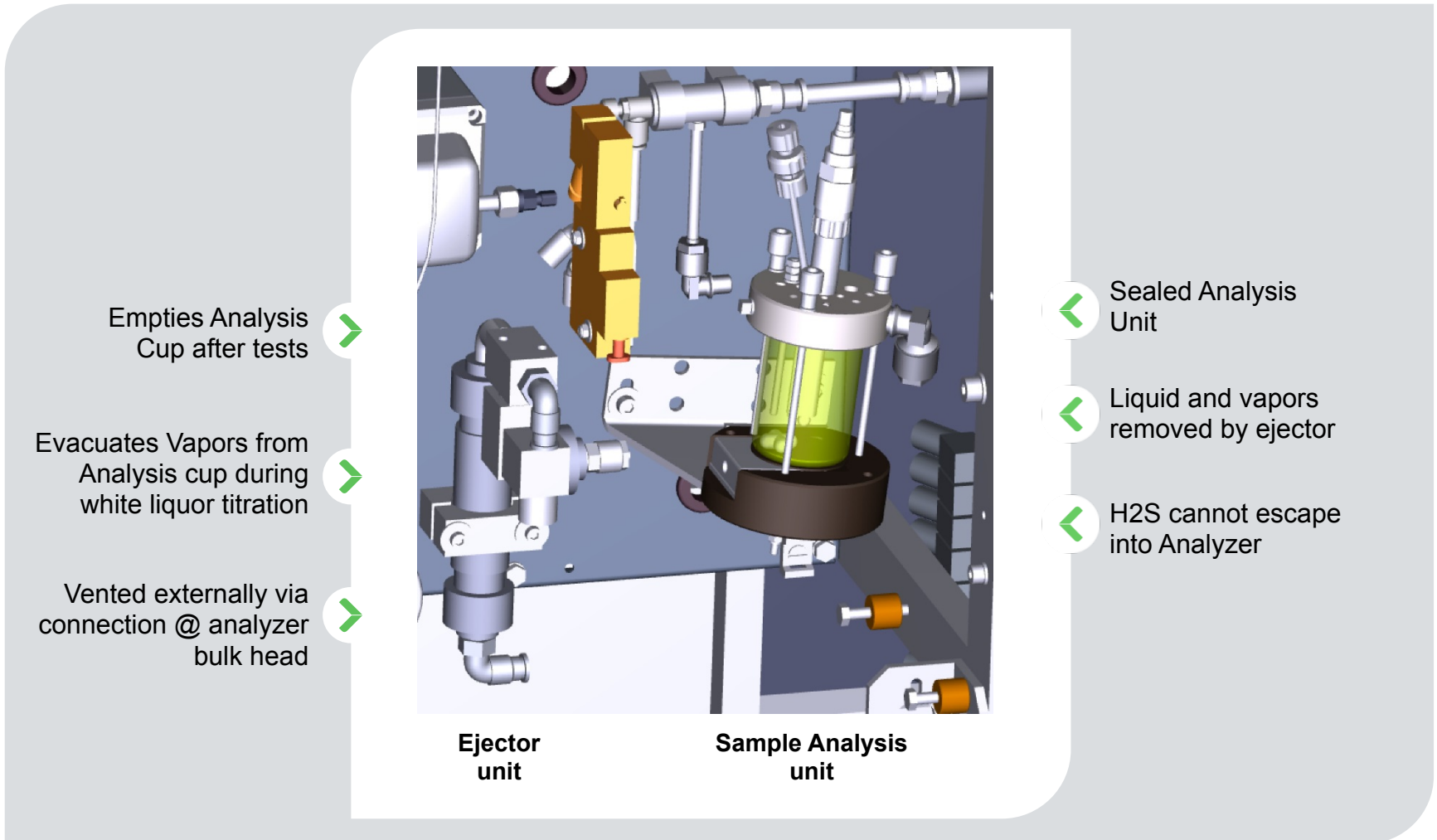
Centralized system

Liquor Titrator Structure



Cooking Liquore Titrator Analysis Unit

Sample Analysis Waste & H₂S Vapor Removal



Accurate results from day one

Based on reliable measurement

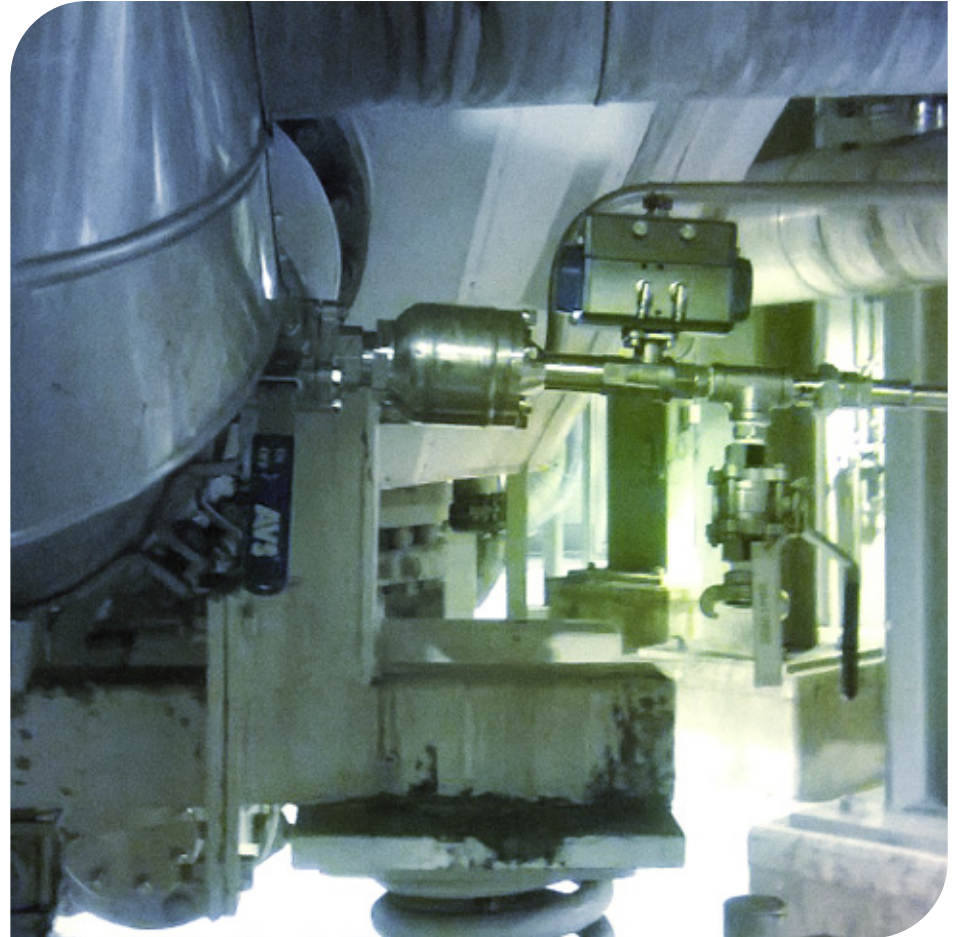
- An effective cooking management tool
- Better pulp quality and production control
- Easy installation without need for calibration
 - Only calibrate new pH probe
- Large reference base
- Field-proven results



Safe and convenient

The multipoint analyzer concept

- Simple installation, cost-effective maintenance
- Centralized location of most critical sample handling and components
- Minimized exposure to equipment in harsh environments
- Automatic self-cleaning system
- Industry-standard communication alternatives and protocols



Automatic Sampling Liquor Titrator

- Alkali Titrator runs the Scandinavian test method in order to perform the ABC titration for white liquor samples
- Black liquor circulations can be sampled automatically also. The titrator can be configured to run a titration to measure the A point, or it can be setup to run to an endpoint pH in order to match the lab unit in obtaining the EA result.
- No calibration is required since it is running the same or very similar test as the lab. This allows the operator or control scheme to utilize the data within a week of start-up.
- The liquor profile across the digester can be stabilized
- Corrections for inaccuracies in the chip feed rate, moisture content, or even poor chip quality can be corrected by measuring the residual alkali as early in the digester as possible.
- The final alkali content can be measured in the wash zone in order to ensure proper residual levels that will protect against lignin re-precipitation and also corrosion issues.

Caustic Plant Liquor Titrator

Online measurements available from day one

- Provides actual liquor chemistry titration results
- No calibration needed – online measurements available from day one
- Measurement Capabilities:
 - Dissolving Tank green liquor
 - Weak Wash
 - Green liquor to slaker
 - Slaker/1st causticizer lime mud slurry
 - Final causticizer lime mud slurry
 - Final white liquor to the digester
 - NEW Recovery Boiler Reduction
- Automatic flushing and acid washing sequences
- Can be used to measure manual liquor samples
- Rugged design ensures high uptime (98%) and low maintenance (clear maintenance program)

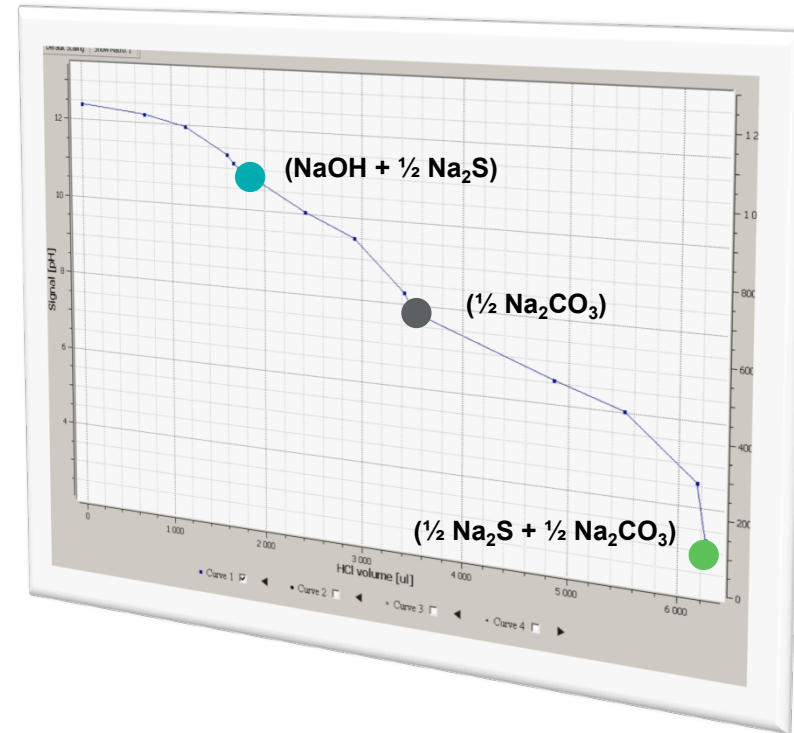


Analysis method

White liquor

- SCAN 30:85
- 2 ml sample is titrated with 1.0 N HCl – an electrode registers pH
- Reactions for the three equivalence points

- **EP1:** $\text{NaOH} + \text{Na}_2\text{S} + \text{Na}_2\text{CO}_3 + 2\text{HCl}$
→ $\text{NaHS} + \text{Na}_2\text{CO}_3 + 2\text{NaCl} + \text{H}_2\text{O}$ pH 10.6
- **EP2:** $\text{NaHS} + \text{Na}_2\text{CO}_3 + \text{HCl}$
→ $\text{NaHS} + \text{NaHCO}_3 + \text{NaCl}$ pH 8.4
- **EP3:** $\text{NaHS} + \text{NaHCO}_3 + 2\text{HCl}$
→ $2\text{NaCl} + \text{H}_2\text{S} + \text{H}_2\text{O} + \text{CO}_2$ pH 3.8



Calculations of Results

Effective Alkali	EA	EA = NaOH + 1/2 Na ₂ S EA = A * NM / V
Active Alkali	AA	AA = NaOH + Na ₂ S AA = (2A - 2B + C) NM / V
Total Titratable Alkali	TTA	TTA = NaOH + Na ₂ S + Na ₂ CO ₃ TTA = C * NM / V
Causticizing Degree	CE%	CE% = NaOH * 100 / (NaOH + Na ₂ CO ₃) CE% = (2 B - C) * 100 / (4B - C - 2A)
Sulfidity	S%	S% = Na ₂ S * 100 / (NaOH + Na ₂ S) S% = (2B - C) * 100 / (2A - 2B + C)
Sodium Carbonate	Na ₂ CO ₃	Na ₂ CO ₃ = (2B - 2A) * NM / V
Sodium Hydroxide	NaOH	NaOH = (2B - C) * NM / V
Sodium Sulfide	Na ₂ S	Na ₂ S = (2A - 4B + 2C) * NM / V

A = Titration volume at equivalence point 1 (EP1)

B = Total titration volume at eq. point 2 (EP2)

C = Total titration volume at eq. point 3 (EP3)

N = HCl normality

V = Sample volume

M = Molecular weight

- if result is computed as Na₂O, M = 31

- if result is computed as NaOH, M = 40

Caustic Plant Liquor Titrator

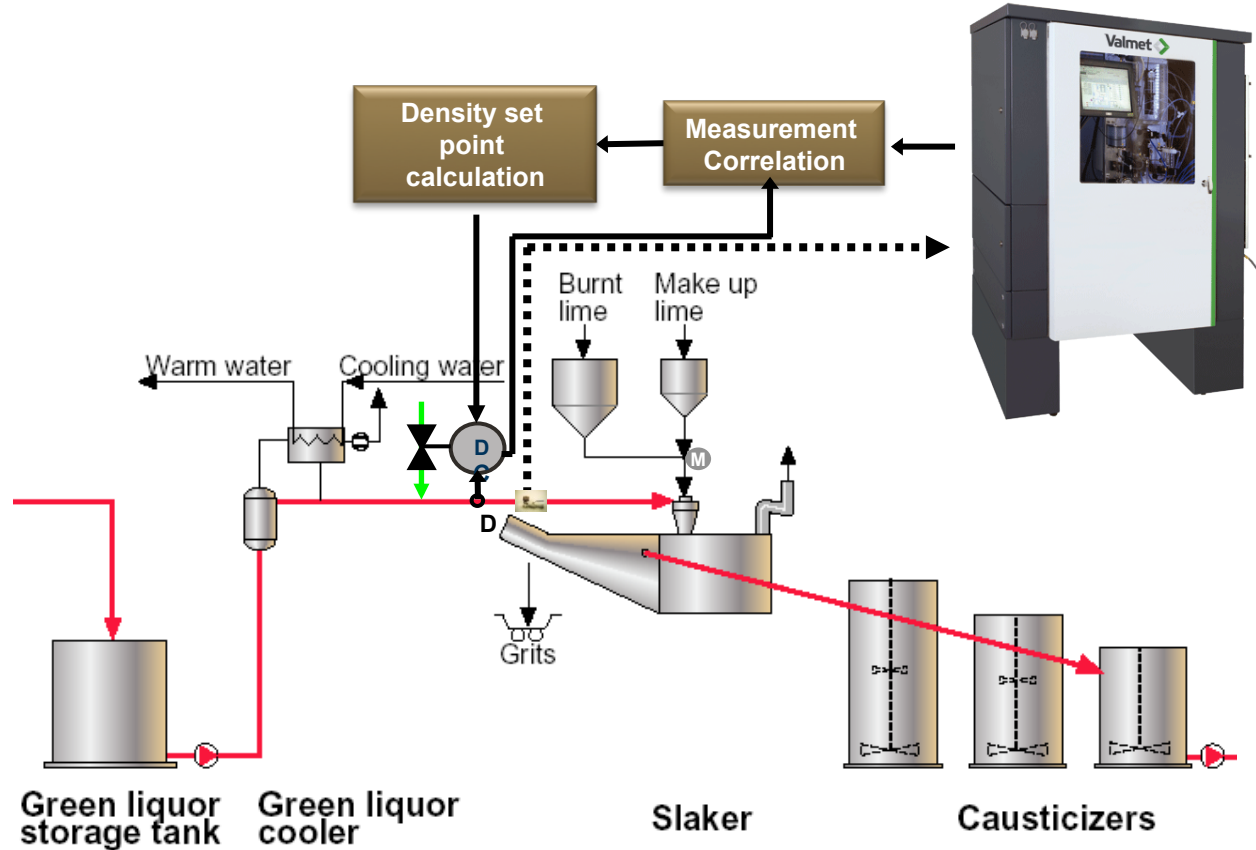
For Employee Safety!

- With automated liquor sample collection preparation and testing, operators and laboratory personnel are exposed less to the safety hazards of manual sample collection
- Eliminate need for operators or laboratory personnel to collect dissolving tank/caustic plant samples for TTA/Density or smelt spout sampling for reduction.
- Automated liquor sampling, sample preparation and testing allows time for operators and laboratory personnel to:
 - Work Safer!
 - Focus on managing the process
 - Improve housekeeping in process areas
 - Assume additional operational tasks
 - Perform more value added tests

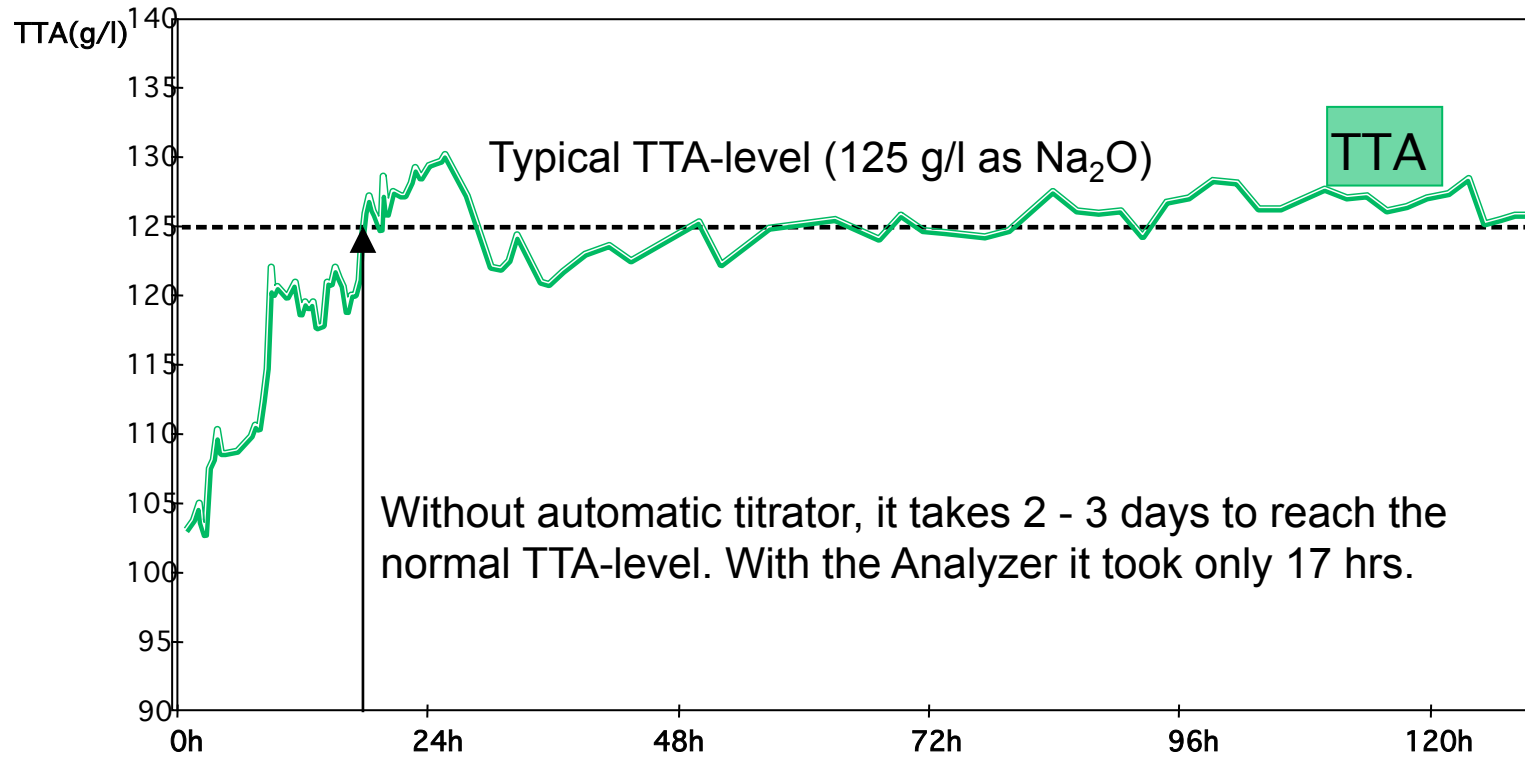


Green Liquor TTA Control - Slaker

- Requires Green Liquor Density Control at the slaker
- Goal is a stable and consistent green liquor TTA to the slaker to minimize variability and maximize lime addition
- Manipulated Variables
 - Weak Wash Flow
- Control Variables
 - Green Liquor Density
 - Green Liquor TTA



Causticizing Process Start-up



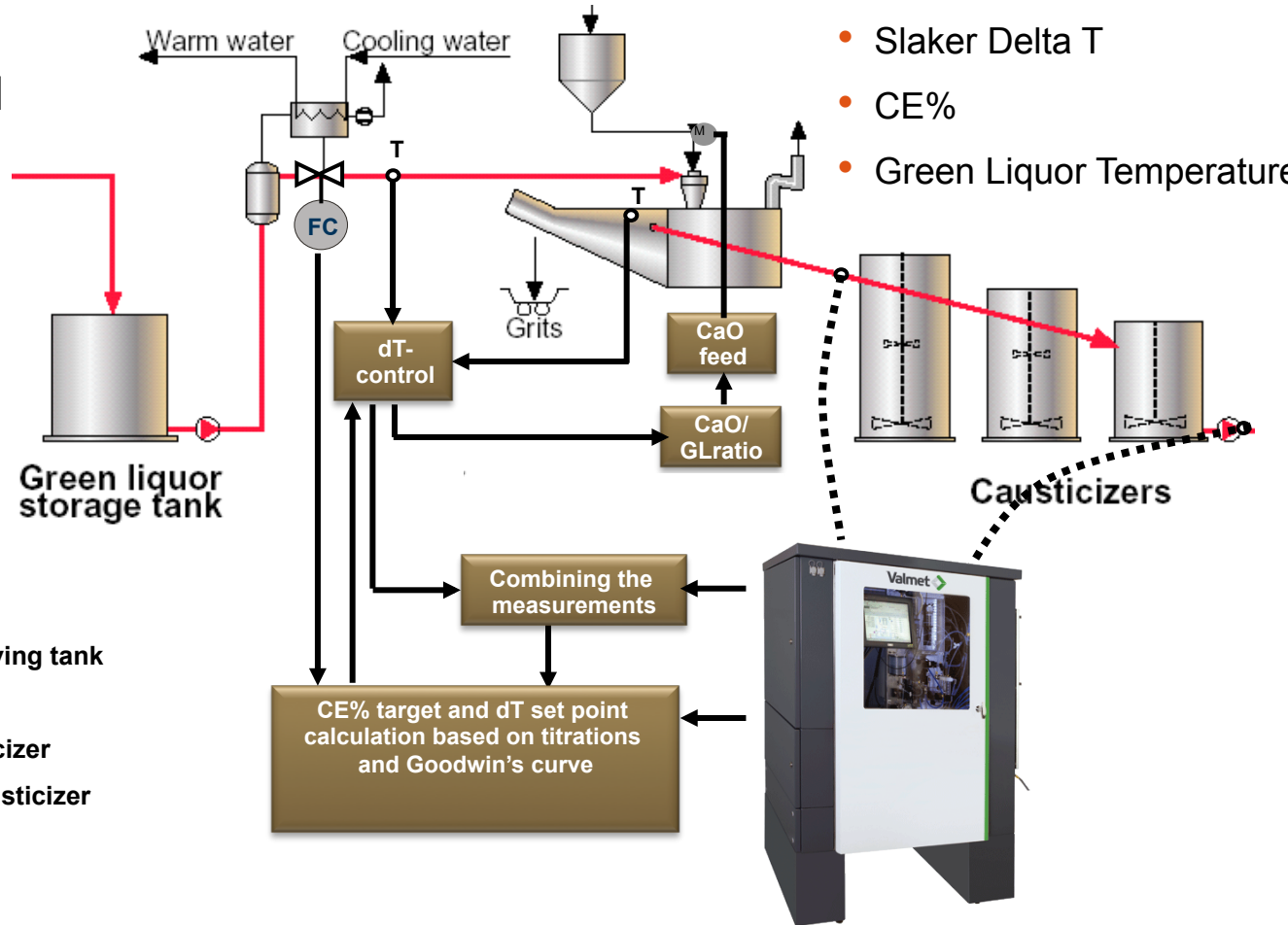
Causticizing Optimization %CE Control

- Incoming TTA Sets the Correct %CE Target Based on Goodwin's Curve
- Slaker Delta T Is Used for Quick Feedback to Lime Reactivity and is Modeled to the %CE
- Green Liquor Temperature can Also Be Controlled to Maximize the Slaker Temp Short of Boiling

Sampling points:

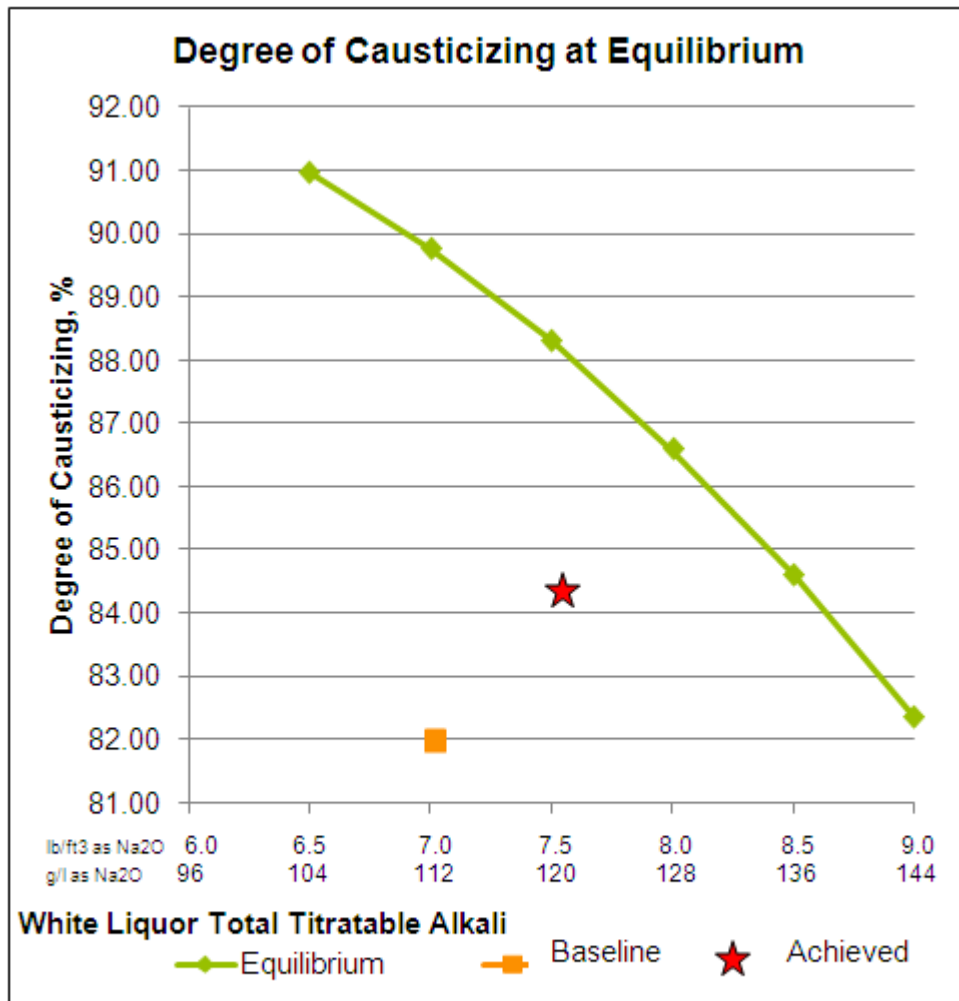
- 1 Green Liquor at dissolving tank
- 2 Green liquor to slaker
- 3 Lime milk at 1st causticizer
- 4 Lime milk after last causticizer
- 5 White liquor CE%

CE% Controls



- Manipulated Variables
 - Lime/Green Liquor Ratio
- Controlled Variable
 - Slaker Delta T
 - CE%
 - Green Liquor Temperature

Causticizing Optimization Results



Mill Base line:

82%

Achieved CE:

84.5%

